

## 12 mm DELRIN NON-STERILIZABLE POLAROGRAPHIC OXYGEN ELECTRODE

### GENERAL INFORMATION

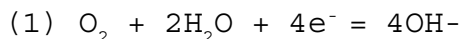
This family of dissolved oxygen electrodes is of the polarographic type, designed to be interchangeable with NBS oxygen amplifiers and other manufacturer's amplifiers designed to the same specifications. The probe is not designed to withstand severe conditions, like high temperature and high pressure.

### SPECIFICATIONS

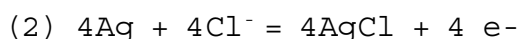
Electrode body	: Black Delrin
Membrane	: Teflon
Cathode	: Platinum, or Gold (varies, depending on order)
Anode	: Tubular Ag (unless otherwise specified)
Output in air-saturated DI water	: 35-60 nA
Residual Signal	: less than 1 nA
Response time	: less than 60 seconds to 98% of full response

### THEORY OF OPERATION

The polarographic dissolved oxygen electrode was first described by Clark in 1956. It consists of polarized platinum and silver electrodes, with the electrolyte separated from the sample by teflon gas permeable membrane. Oxygen diffuses across the electrode membrane and is reduced to hydroxyl ions at the platinum cathode according to the reaction;



The electrons necessary for this process are produced by a reaction at the silver anode. Because the electrolyte contains chloride ions, this reaction occurs as;



At any given temperature, the current flow between cathode and anode is directly proportional to the level of oxygen outside the membrane.

### **ZERO POINT OF AMPLIFIER**

The electrical zero point of the amplifier must be first set. (Follow the manufacturer's operating instructions).

### **POLARIZATION**

When the amplifier is switched on, polarization voltage is applied to the anode and the cathode. Initial current is very strong, but falls off exponentially and settles down to a steady state after a few hours.

Since the equilibrium period of the polarization current is relatively long, it is advisable to keep the electrode connected to the amplifier when not in use. The small current flowing through the electrode will not shorten the life of the electrode. If, for any reason, the electrode must be disconnected, or the amplifier switched off, the electrode must be re-polarized before it can be used. During the polarization period, the electrode current will decrease even in oxygen free solutions. An excessive zero current may indicate incomplete polarization.

### **ELECTRODE CONSTRUCTION**

The external portion of the electrode is constructed of a black delrin cylinder. The functional portion of the electrode is constructed of a ryton chamber enclosing an ultra-pure tubular Ag anode and a platinum cathode. A thin teflon membrane at the tip provides for efficient sealing of the system and is in contact with the platinum cathode and electrolyte. The membrane is permeable to oxygen but impermeable to water and electrolyte.

### **PREPARATION**

The electrode is shipped filled with electrolyte and with the teflon membrane in place. The protective vinyl cap on the sensing tip must be removed before and during operation. Upon receiving the electrode, inspect for damage. If any damage is noticed, notify pHoenix Electrode Company Service Department immediately.

## **ELECTRODE CHECK**

Remove protective vinyl cap.

Connect the detachable electrode cable to the probe and amplifier.

Follow polarization procedures (see **POLARIZATION SECTION**)

Leave electrode in air saturated DI water. Set the range selector on the amplifier to zero. Adjust the zero current to read zero on the indicator. Then calibrate the meter to 100% saturation degree. Pass nitrogen gas over the electrode membrane. The electrode will react immediately increasing the reading, which will decrease after 45 seconds. The reading should be under 10%. (Refer to the amplifier manufacturer's procedure on checking the electrode).

## **ZERO POINT OF THE ELECTRODE**

The zero current of the dissolved oxygen electrode (electrode current in pure  $N_2$ ) is usually negligibly small and almost identical with the amplifier zero point. Nonetheless, the electrode zero point should be periodically checked as some electrode faults result in excessive zero current. Moreover, checking the zero point calibration is necessary before measurement of low oxygen concentration. The zero current (residual current) of the electrode may be compensated by adjusting the zero button on the meter. After zeroing the electrode, recalibrate it in air-saturated water.

## **ELECTRODE SLOPE**

Zero point adjustment must precede slope calibration. Unlike zero point calibration, the aqueous phase is preferred for slope adjustment. The following problems arise in calibration in air:

- 1) Membrane permeability slightly differs in air and water.
- 2) Relative air humidity rarely attains 100%.
- 3) The temperature is poorly defined.

Calibration is usually most effective at an air saturation of 100%. Since calibration depends on pressure, the 100% air saturation should be done under operating pressure. (Refer to amplifier manufacturer's operating procedures).

## **CALIBRATION**

Connect the electrode cable to the DO probe and to the DO amplifier. Follow the polarization procedure as described, along with the amplifier manufacturer's operating manual.

Dissolved oxygen electrodes should be recalibrated prior to every measurement.

## **MEMBRANE REPLACEMENT**

The membrane should be examined routinely after each use and replaced if any deterioration is evident.

To remove the membrane, unscrew the bottom cap. Remove guarded portion from cap assembly and discard old membrane.

Inspect the glass part and platinum cathode at the very tip of the glass and wipe with tissue paper.

Remove a new membrane from its package. Lay new membrane across cap and press guarded portion onto the cap assembly.

Fill the membrane cap with electrolyte and screw the cap onto the tip of the electrode.

After the membrane cap is securely in place, the teflon membrane should be stretched across the glass cathode. The electrode is to be filled with electrolyte, following the above procedures.

After re-assembling the membrane and filling the electrode with electrolyte, it must be checked. The polarization procedures should be followed.

## **TROUBLESHOOTING**

Problem: The probe when in air-saturated water generates no potential (zero output).

Solution: Check that all signal cables connections are correct. Check that the probe is filled with electrolyte.

Problem: The output signal is reversed.

Solution: Reverse the signal wires connected to the display module or recorder.

Problem: The calibration potentiometer is on maximum but the signal does not reach 100%.

Solution: Change the membrane. Change the electrolyte.

Problem: The output signal is unstable and shows random drift.

Solution: Check the grounding of the fermentor and of the instrumentation.

**NOTE:** The output signal may not reach 100% if the ambient air temperature is less than 15%.

### **MAINTENANCE**

After the electrode has been used for a period of time the residual current may rise. To minimize the residual current, the following cleaning method is recommended. Gently unscrew the membrane cover from the body, rinse the inside of the membrane with DI water. Flush the electrolyte chamber with DI water. Soak the silver anode and the glass tip in silver cleaning solution (commercially available as jewelry cleaning solution) for about 10 minutes. Thoroughly rinse the soaked portion with DI water, blot dry with tissue paper. Fill the electrode with electrolyte as described above.

### **PACKING LIST**

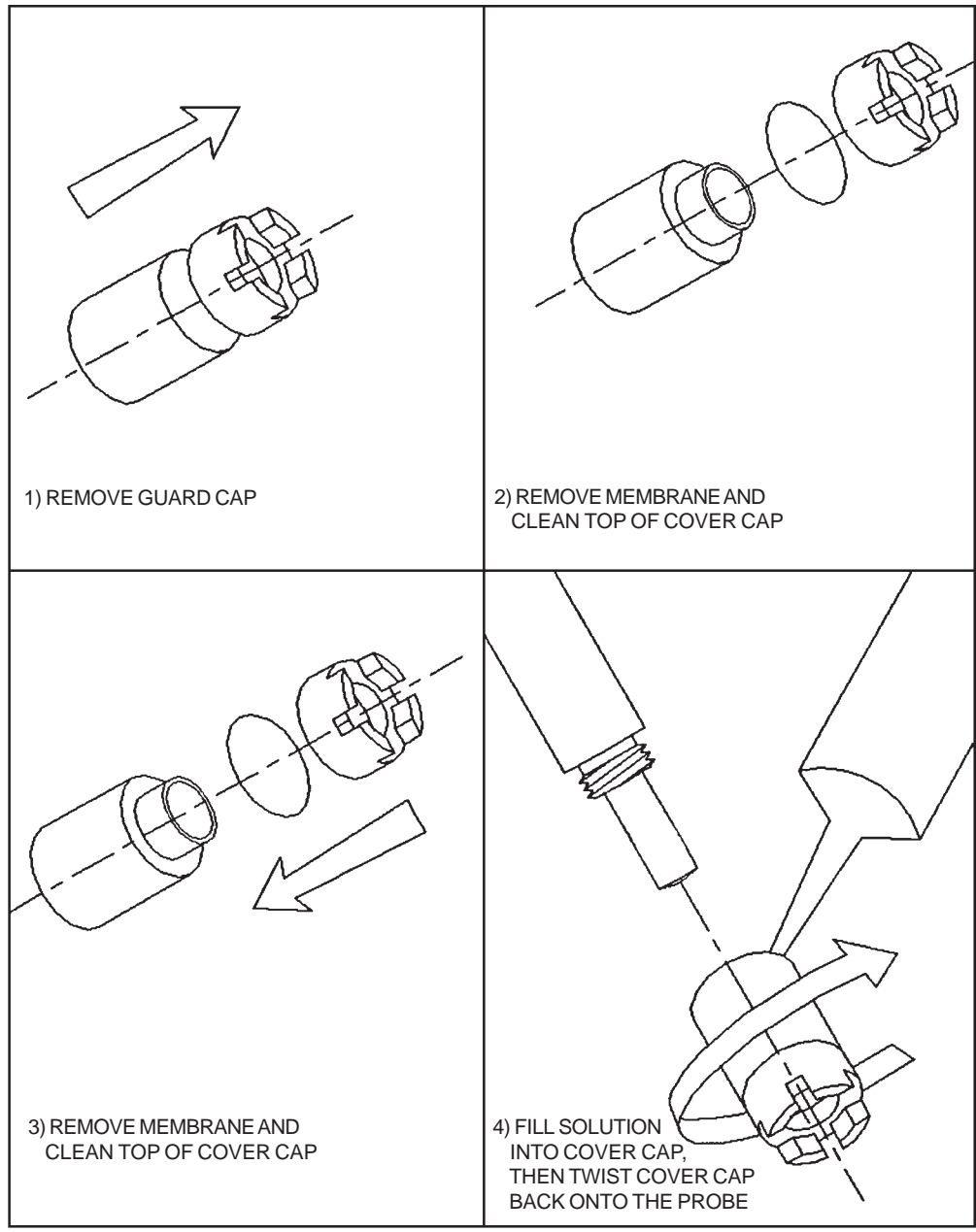
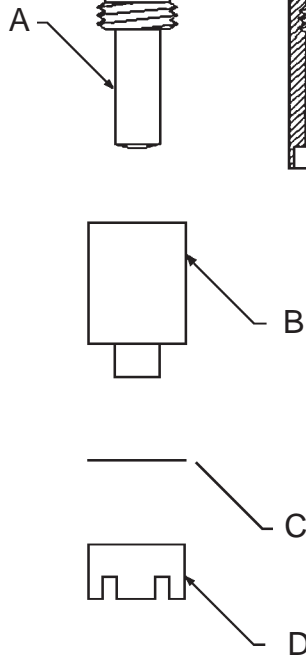
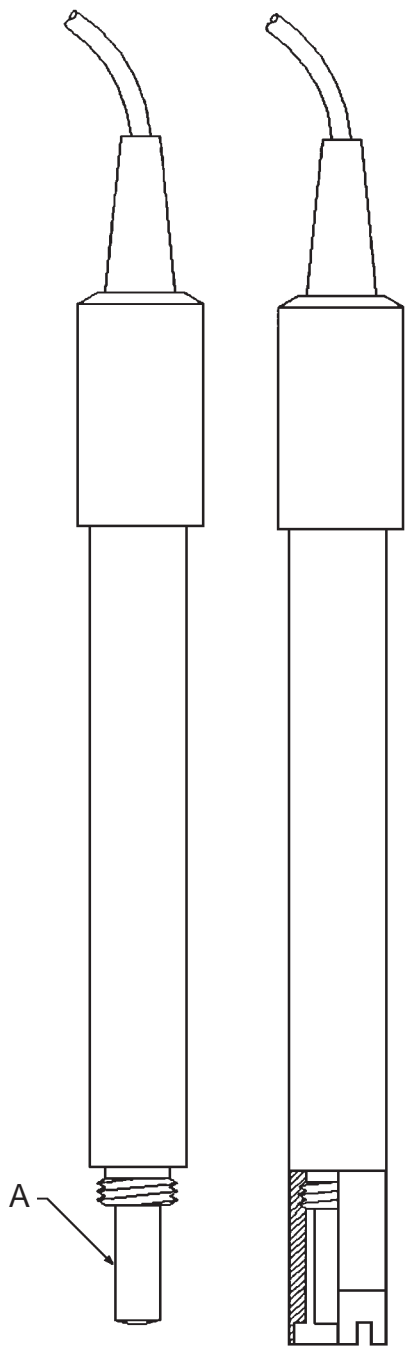
pHoenix Electrode Company polarographic DO probe

Electrolyte 1

Spare membranes 1

O-rings 1

Plastic syringe 1



1) REMOVE GUARD CAP

2) REMOVE MEMBRANE AND CLEAN TOP OF COVER CAP

3) REMOVE MEMBRANE AND CLEAN TOP OF COVER CAP

4) FILL SOLUTION INTO COVER CAP, THEN TWIST COVER CAP BACK ONTO THE PROBE

- A SILVER (ANODE)
- B COVER CAP
- C MEMBRANE
- D GUARD CAP

## POLAROGRAPHIC